

Question 1

Answer

Correct

Question 1: Identify the correct chemical species and explain why/where they are present in the system. (10 marks)

Reaction: $2\text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$

Species: H_2O , H_3O^+ , OH^-

Reaction: $\text{H}_2\text{O} + \text{CO}_2 \rightleftharpoons \text{H}_2\text{CO}_3$

Species: H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-}

Reaction: $\text{H}_2\text{O} + \text{NH}_3 \rightleftharpoons \text{NH}_4^+ + \text{OH}^-$

Species: H_2O , NH_3 , NH_4^+ , OH^-

Reaction: $\text{H}_2\text{O} + \text{H}_2\text{S} \rightleftharpoons \text{HS}^- + \text{H}_3\text{O}^+$

Species: H_2O , H_2S , HS^- , H_3O^+

Reaction: $\text{H}_2\text{O} + \text{H}_2\text{PO}_4^- \rightleftharpoons \text{H}_2\text{PO}_4^- + \text{H}_3\text{O}^+$

Species: H_2O , H_2PO_4^- , H_3O^+

Reaction: $\text{H}_2\text{O} + \text{HPO}_4^{2-} \rightleftharpoons \text{HPO}_4^{2-} + \text{H}_3\text{O}^+$

Species: H_2O , HPO_4^{2-} , H_3O^+

Reaction: $\text{H}_2\text{O} + \text{H}_2\text{N}_2\text{O}_2 \rightleftharpoons \text{H}_2\text{N}_2\text{O}_2 + \text{H}_3\text{O}^+$

Species: H_2O , $\text{H}_2\text{N}_2\text{O}_2$, H_3O^+

Question 2: Write the chemical formulae for the species that you think are present in the system. (10 marks)

Reaction: $\text{H}_2\text{O} + \text{CO}_2 \rightleftharpoons \text{H}_2\text{CO}_3$

- (a) H_2O , CO_2 , H_2CO_3
- (b) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-}
- (c) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+
- (d) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^-
- (e) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^-
- (f) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^- , HPO_4^{2-}
- (g) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^- , HPO_4^{2-} , PO_4^{3-}
- (h) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^- , HPO_4^{2-} , PO_4^{3-} , $\text{H}_2\text{N}_2\text{O}_2$
- (i) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^- , HPO_4^{2-} , PO_4^{3-} , $\text{H}_2\text{N}_2\text{O}_2$, H_2S
- (j) H_2O , CO_2 , H_2CO_3 , HCO_3^- , CO_3^{2-} , H_3O^+ , OH^- , H_2PO_4^- , HPO_4^{2-} , PO_4^{3-} , $\text{H}_2\text{N}_2\text{O}_2$, H_2S , HS^-

Question 3: Write the chemical formulae for the species that you think are present in the system. (10 marks)

Reaction: $\text{H}_2\text{O} + \text{H}_2\text{S} \rightleftharpoons \text{HS}^- + \text{H}_3\text{O}^+$

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